

# EILERS GRADUATE STUDENT FELLOWSHIP FINAL REPORT

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<b>FACULTY ADVISOR:</b>	Casey P. O'Brien
<b>REPORT PERIOD:</b>	2022
<b>PROJECT TITLE:</b>	Operando Surface Enhanced Raman Spectroscopy (SERS) Platform for Studying the Structure and Dynamics of Amine-based Membranes in Complex Environments
<b>CONNECTION TO ND ENERGY'S RESEARCH AREAS (CHECK ALL THAT APPLY):</b>	<input type="checkbox"/> Energy Conversion and Efficiency <input type="checkbox"/> Sustainable and Secure Nuclear <input type="checkbox"/> Smart Storage and Distribution <input type="checkbox"/> Transformation Solar <input checked="" type="checkbox"/> Sustainable Bio/Fossil Fuels <input type="checkbox"/> Transformative Wind

## MAJOR GOALS AND ACCOMPLISHMENTS:

List your major research goals and provide a brief description of your accomplishments (1-2 sentences). Indicate the percentage completed for each goal. Please use a separate sheet to share additional details, technical results, charts, and graphics.

MAJOR RESEARCH GOALS	ACTUAL PERFORMANCE AND ACCOMPLISHMENTS	% OF GOAL COMPLETED
Design, develop, and test a new experimental <i>operando</i> Raman spectroscopic platform that could probe the chemical structure and dynamics of membranes in realistic operating environments.	A new experimental <i>operando</i> surface-enhanced Raman spectroscopic (SERS) platform has been established, which can characterize the molecular structure of polymeric membranes and simultaneously measure their gas separation performance under realistic operating conditions. The transport intermediates of CO <sub>2</sub> in PVAm membrane were successfully demonstrated using the new <i>operando</i> platform.	100
Model study of Amine-Containing CO <sub>2</sub> -Selective Membrane Process for Carbon Capture.	A mathematical model was proposed to describe the transport of CO <sub>2</sub> through the PVAm membrane. The model quantitatively describes the permeability of CO <sub>2</sub> via facilitated transport is an order of magnitude higher than that via solution diffusion. Based on the <i>operando</i> SERS platform, the effect of CO <sub>2</sub> partial pressure on CO <sub>2</sub> transport was also studied in molecular scale.	100
Study and address the knowledge gap between different amine-based polymeric membrane structures (PVAm, PMVAm, PDVAm, and P4VP) and performance.	Different amine-based membranes were investigated with the <i>operando</i> SERS platform to understand the structure effect on CO <sub>2</sub> transport mechanism, and to understand which amine types and operating conditions exhibit the optimal CO <sub>2</sub> transport performance.	60

## RESEARCH OUTPUT:

Please provide detailed information below regarding any output resulting from your research project.

CATEGORY	INFORMATION
<b>EXTERNAL PROPOSALS</b>	National Science Foundation CAREER (CBET-2144362), "Catalytic Membranes for Integrated CO <sub>2</sub> Capture and Conversion," 5 years (June 2022 – May 2027) \$537,317.
<b>EXTERNAL AWARDS</b>	N.A.
<b>JOURNAL ARTICLES</b>	<ol style="list-style-type: none"> <li>ACS Applied Materials &amp; Interfaces; <i>Operando</i> Surface-Enhanced Raman-Scattering (SERS) for Probing CO<sub>2</sub> Facilitated Transport Mechanisms of Amine-Functionalized Polymeric Membranes; <b>Hui Xu</b>, Justin Easa, Sarah G. Pate, Renxi Jin, and Casey P. O'Brien; February 14, 2022; March 22, 2022, 14, 15697-15705.</li> <li><i>In Submission</i>, Mathematical Modeling of CO<sub>2</sub> Facilitated Transport across Polyvinylamine Membranes with Direct <i>Operando</i> Observation of Amine Carrier Saturation; <b>Hui Xu</b>, Sarah G. Pate, and Casey P. O'Brien.</li> </ol>
<b>BOOKS AND CHAPTERS</b>	
<b>PUBLIC PRESENTATIONS, SEMINARS, LECTURES</b>	ND Energy Postdoc and Grad Student Seminar, " <i>Operando</i> Surface-Enhanced Raman-

	Scattering (SERS) Platform for Studying the Structure and Dynamics of CO <sub>2</sub> Facilitated Transport in Amine-based Membranes in Complex Environments”, November 2022, Notre Dame
<b>AWARDS, PRIZES, RECOGNITIONS</b>	N.A.
<b>INTERNAL COLLABORATIONS FOSTERED</b>	Collaborated with Professor William Philip (Chemical and Biomolecular Engineering) and Professor Jon Camden (Chemistry and Biochemistry).
<b>EXTERNAL COLLABORATIONS FOSTERED</b>	N.A.
<b>WEBSITE(S) FEATURING RESEARCH PROJECT</b>	N.A.
<b>OTHER PRODUCTS AND SERVICES</b> (e.g., media reports, databases, software, models, curricula, instruments, education programs, outreach for ND Energy and other groups)	N.A.

### **MAJOR GOALS AND ACCOMPLISHMENTS** (Additional Details, Technical Results, Charts and Graphics)

**Goal 1:** Design, develop, and test a new experimental *operando* Raman spectroscopic platform that could probe the chemical structure and dynamics of membranes in realistic operating environments. This was done. The new experimental *operando* surface-enhanced Raman spectroscopic (SERS) platform has been successfully established (Fig. 1), which can characterize the molecular structure of polymeric membranes and simultaneously measure their gas separation performance under realistic operating conditions. Polyvinylamine (PVAm) membrane was prepared and used to investigate the utility of the new *operando* SERS platform (Fig. 2). The experiments showed that a small amount of plasmonic Ag nanoparticles added to the PVAm solution prior to knife-casting onto PVDF supports selectively enhances the Raman signal intensity associated with CO<sub>2</sub> transport intermediates (carbamate) formed in PVAm during the separation of CO<sub>2</sub>/CH<sub>4</sub> (Fig. 2b). The intensity of the Raman bands associated with carbamate species correlates with the CO<sub>2</sub> flux across the membrane (Fig. 2d), indicating that these are the primary intermediates in the CO<sub>2</sub> transport mechanism across PVAm. The new *operando* spectroscopy platform can be applied to other types of membranes, such as mixed-matrix membranes and ceramic membranes, to provide detailed information under realistic working circumstances, which could help to understand the key factors that determine membrane performance and design new membranes with enhanced performance.

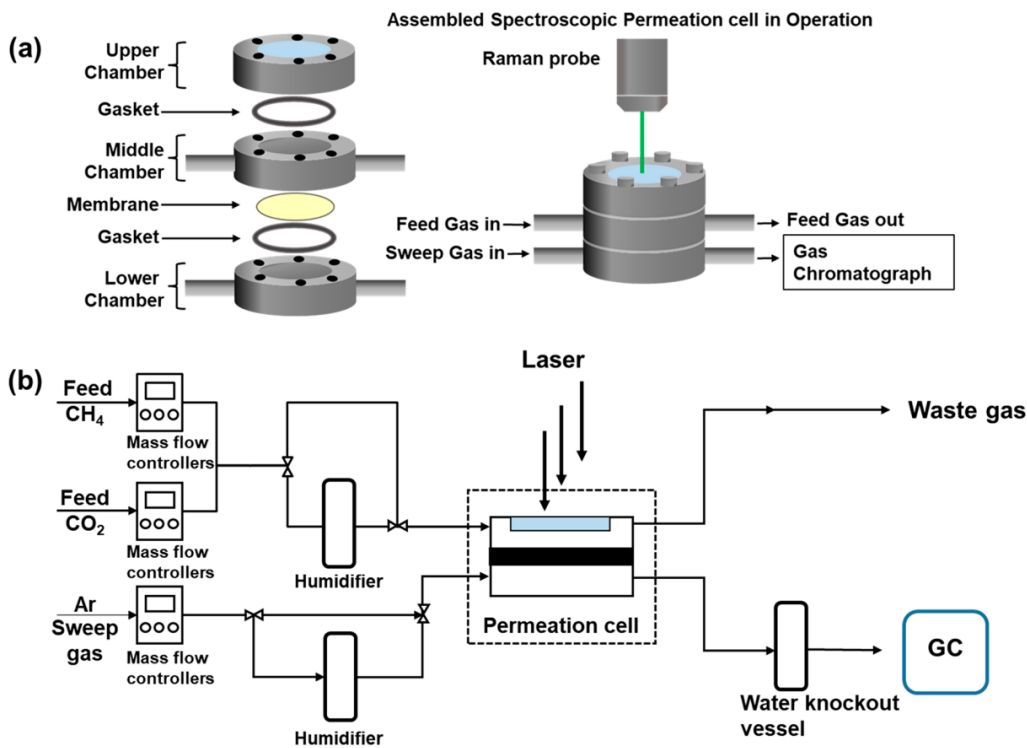


Fig. 1. Schematic illustration of (a) Permeation Cell Exploded (left) and Assembled (right) for Gas Separation Performance Evaluation and (b) the Permeation Apparatus

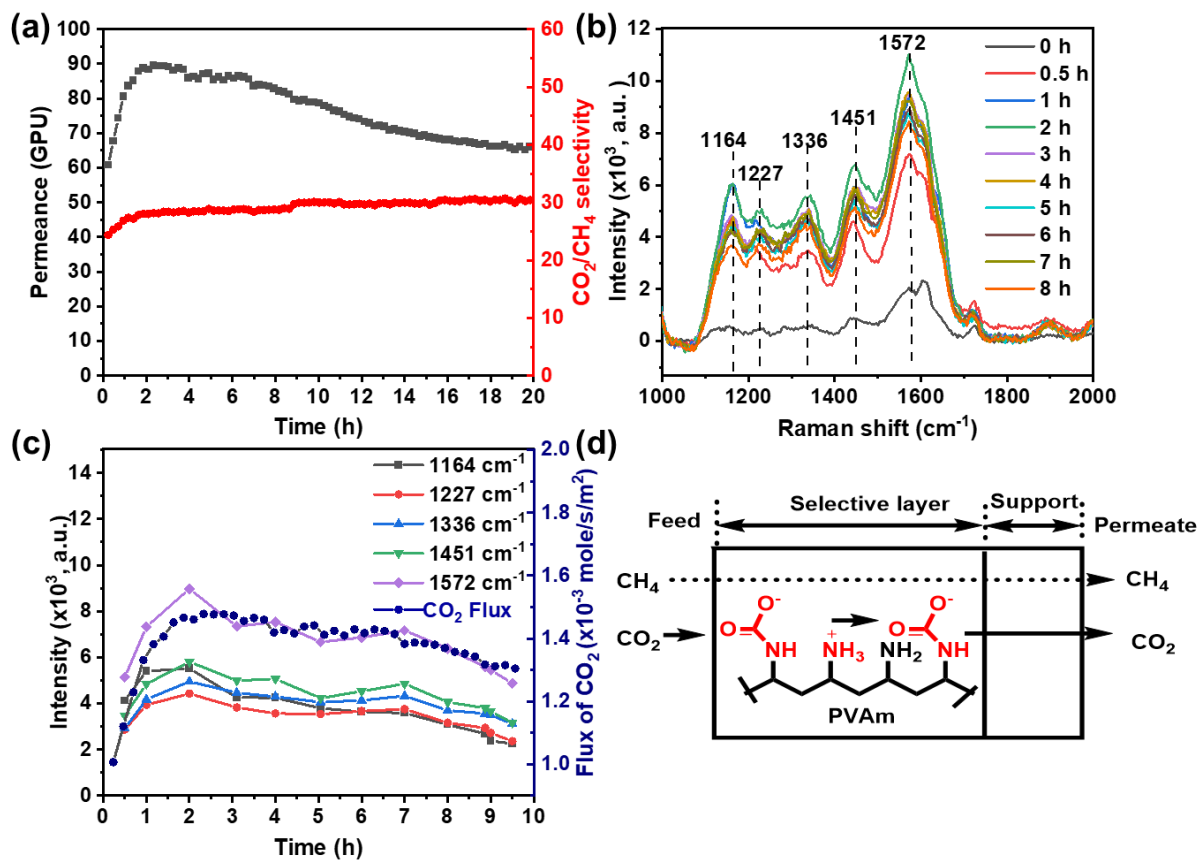


Fig. 2. (a) CO<sub>2</sub> transport performance of PVAm/Ag NPs(20.4mg)-PVDF membrane (PVDF acts as the substrate, while PVAm and Ag nanoparticles form the active separation membrane together); (b) In-situ Raman spectra as a function of exposure time; (c) Chemical structure of PVAm and facilitated CO<sub>2</sub> transport mechanism; (d) Intensity changes of peaks over exposure time and flux of CO<sub>2</sub> over exposure time.

**Goal 2:** Model study of Amine-containing CO<sub>2</sub>-selective Membrane Process for Carbon Capture. This was done. As the CO<sub>2</sub> permeance of facilitated transport membrane is pressure-dependent, the effect of CO<sub>2</sub> partial pressure on CO<sub>2</sub> facilitated transport across PVAm membrane was also studied with our apparatus to study the trend and correlate the phenomenon with molecular- and nano-scale structure. A carrier saturation phenomenon was observed, and a mathematical transport model for CO<sub>2</sub> transport through the amine-containing FTMs was developed with fitting parameters that are verifiable by experiment data (Fig 3). The model quantitatively describes the permeability of CO<sub>2</sub> via facilitated transport is an order of magnitude higher than that via solution diffusion. Based on the operando SERS platform, the effect of CO<sub>2</sub> partial pressure on CO<sub>2</sub> transport was also studied on the molecular scale, which provided direct spectroscopic evidence of the carrier saturation phenomenon (Fig. 4). Our modeling and spectroscopic approaches are not limited to this particular membrane system and may be used to study other amine-based FTMs for CO<sub>2</sub> capture or FTMs for other types of separations such as olefin-paraffin separation, for example.

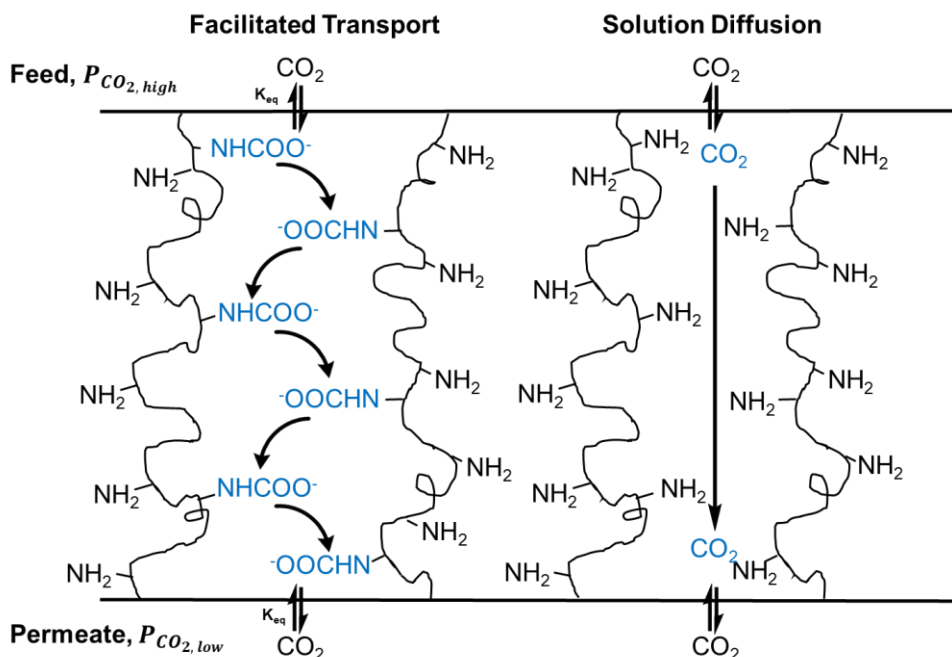


Fig. 3. Schematic representation of CO<sub>2</sub> transport across PVAm via facilitated transport of carbamate and solution-diffusion of CO<sub>2</sub>.

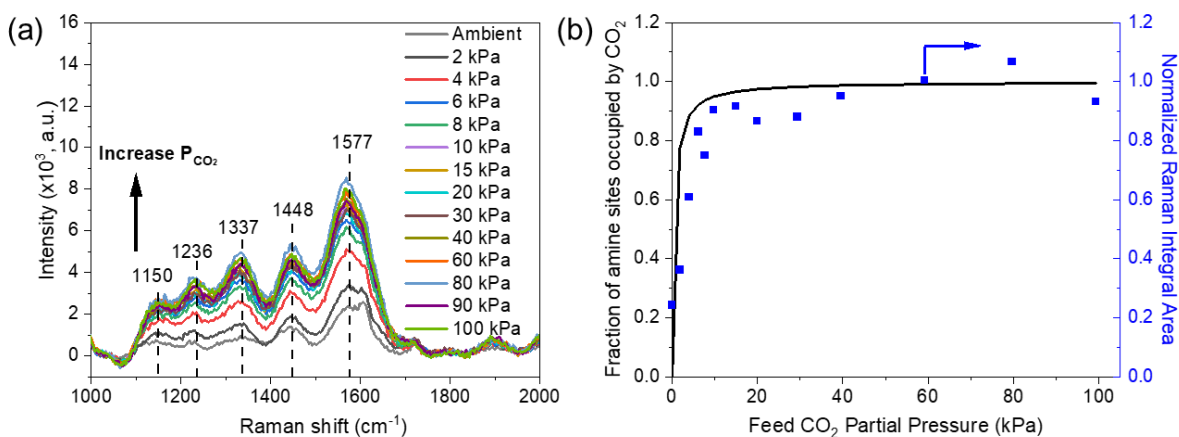


Fig. 4. (a) *Operando* Raman spectra as a function of CO<sub>2</sub> partial pressure during exposure to humidified CO<sub>2</sub>/CH<sub>4</sub> with 2-99 kPa CO<sub>2</sub>, 2.2 kPa H<sub>2</sub>O, and balance CH<sub>4</sub> at atmospheric pressure and room temperature. (b) Fraction of amine sites occupied by CO<sub>2</sub> on the feed side of PVAm from model predictions (black line), and the normalized integral area of the Raman peaks between 1076-1694 cm<sup>-1</sup> (blue squares), versus feed CO<sub>2</sub> partial pressure.

**Goal 3:** Use the new operando Raman spectroscopic platform to study and address the knowledge gap between different amine-based polymeric membrane structures (PVAm, PMVAm, PDVAm, and P4VP) and performance. Other amine-based membranes were investigated further to understand the structural effect of membrane materials on the CO<sub>2</sub> transport mechanism and which amine types and operating conditions exhibit optimal CO<sub>2</sub> transport performance. The SERS-active polymeric membranes were prepared and tested with our operando SERS technique. According to the result, new peaks appeared during their separation process. Comparing the CO<sub>2</sub> separation performances, we found that involving Ag NPs doesn't have much effect on the selectivity of the membranes, although there are changes in their CO<sub>2</sub> permeance. The Raman peaks assignment and study of the role of water in CO<sub>2</sub> transport through the membranes still need to be finished.

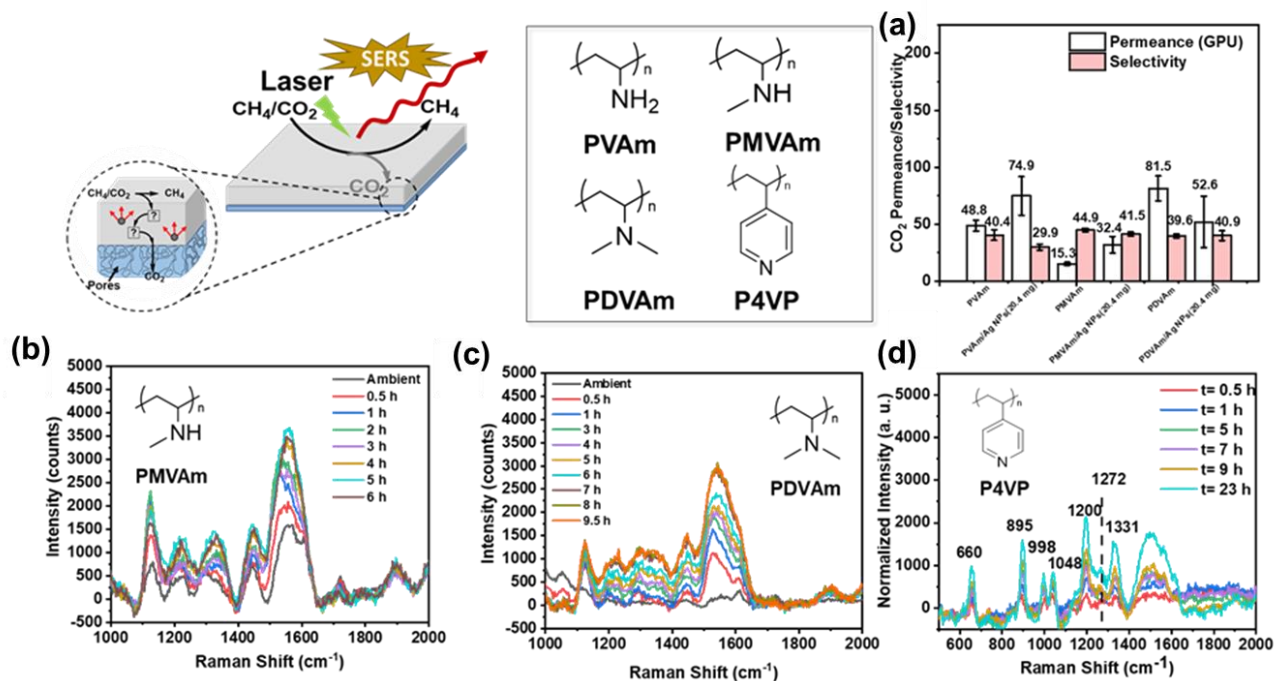


Fig. 5. (a) Comparison of gas separation performance of the membranes with and without Ag NPs involving when separate the simulated biogas. (b) In-situ Raman spectra of PMVAm membrane. (c) In-situ Raman spectra of PDVAm membrane. (d) In-situ Raman spectra of P4VP membrane (feed side: 49 mol% CH<sub>4</sub>, 49 mol% CO<sub>2</sub>, 2 mol% H<sub>2</sub>O, 1atm; sweep side: 98 mol% Ar, 2 mol% H<sub>2</sub>O, 1 atm; Temperature: 19 °C).