

**Fall 2016 Lecture Series**  
**Chemical & Biomolecular Engineering**

**Gold Catalysts for Hydrogen Purification: Mechanistic Tools,  
Hammett Studies, and a Unifying Mechanism for  
CO Oxidation over Au**



**Bert Chandler**  
Professor  
Department of Chemistry  
Trinity University

Hydrogen, the smallest molecule, is a critical industrial chemical for fertilizer (and therefore food) production as well as transportation fuels. Global industrial hydrogen production is a massive industry, totaling over 50 million tons annually and accounting for roughly 3% of global annual energy use. Hydrogen is typically produced via the steam reforming of natural gas, followed by water-gas shift units; thermodynamic and kinetic limits for these processes generally result in a hydrogen feed that contains about 1% CO, which must be removed before it is used in downstream catalysts. Current hydrogen purification methods (methanation or pressure swing adsorption) either consume substantial amounts of hydrogen or are relatively slow and capital intensive. The CO PROX reaction, which preferential oxidizes CO with O<sub>2</sub> in the H<sub>2</sub> feed, is a promising alternative to these methods.

Supported gold nanoparticles are now well known to be the most active low temperature CO oxidation catalysts available. Gold is also a notoriously bad hydrogenation catalyst, so it ought to be ideal for PROX; however, despite intense study over the last 15 years, there have been exceedingly few reports of successful PROX catalysts. This seminar will focus on our efforts to develop kinetic tools to better understand supported gold catalysts, build a mechanistic understanding of the CO oxidation reaction, and our early successes in CO PROX. We are applying a number of mechanistic chemistry techniques, such as Hammett studies, kinetic isotope effects, and Michaelis-Menten kinetics, to better understand how support effects and water tune the chemistry of Au nanoparticle surfaces. These studies, combined with collaborative density functional theory calculations, have led us to a deeper understanding of O<sub>2</sub> activation over Au, a new proposed mechanism for CO oxidation, and significant advances in CO PROX catalysis.

**Tuesday, October 25, 2016, 12:30 pm**

**155 DeBartolo Hall**

**[CBE.ND.EDU/SEMINARS](http://CBE.ND.EDU/SEMINARS)**